

Shri Shivaji Education Society's Board for Higher Education Vidyanagar Karad





YASHWANTRAO CHAVAN COLLEGE OF SCIENCE, KARAD

Department of Chemistry

Year 2019-20

Name of activity- "To determine percentage of nitrogen in the Soil and pH of Soil."

Index

Sr. No.	Name		
1.	Priyanka Vinayak Rokade		
2.	Arjun Jagannath Lohar		
3.	Satish Hanmant Yadav		
4.	Dasharat Antu Yadav		
5.	Jalindar Sadhu Mankar		
6.	Dipak Shivajirao Mali		
7.	Mahesh Pralhad Kale		
8.	Sanjay Sakharam Jadhav		

Head

Department of Chemistry

Yashwantrao Chavan College of

Science, Karad

CHEMISTRY Science

Principal
Yashwantrao Chavan College
of Science, Karad

Yashwantrao Chavan College of Science, Karad

Department of Chemistry Extension Activity 2019-20

Name of activity- "To determine percentage of Nitrogen in the Soil and pH of Soil."

Purpose:	Understanding Soil Dynamics: Gain a comprehensive understanding of the role of nitrogen in soil, its various forms, and the significance of soil pH in influencing plant growth and ecosystem health. Crop Management Strategies: Explore and formulate effective crop management strategies based on nitrogen levels and soil pH, considering the specific needs of different crops. Analytical Proficiency: Develop proficiency in using analytical techniques to accurately determine the percentage of nitrogen in soil samples and to measure soil pH.			
No. of beneficiaries:	08			
Outcome/ success achieved:	Interpretation of Results: Develop the ability to interpret and analyze results obtained from nitrogen determination and pH measurement, considering the implications for soil fertility and plant health. Soil Fertility Assessment: Apply the knowledge gained to assess soil fertility based on nitrogen content and pH levels, contributing to informed agricultural practices. Understanding Nitrogen in Soil: Gain a comprehensive understanding of the role of nitrogen in soil, including its forms, cycling, and impact on plant growth.			
Teachers involved in the activity	Prof. Dr. S. H. Burungale Prof. Dr. A. V. Mali Mr. A. N. Bhingare Dr. R.S. Patil Mr. B. E. Mahadik Mr. G. B. Dhake Dr. U. P. Lad Mr. S. D. Karande Dr. S. D. Jadhav Mrs. P. P. Patil Mrs. S. R. Veer Mrs. M. B. Jagadale			

Head

Department of Chemistry

Yashwantrao Chavan College of

Science, Karad

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Yashwanirao Chavan College of Science, Karad

Department of Chemistry

Name of Activity - To determine percentage of Nitrogen in the Soil pH of the soil.

Sr.				T	
No	Name of Farmer	Address	Soil Received	Mobile No.	Sign
			Date		
1)	रोकडे प्रियोका विनायक	वाळवा संश्राती	7-8-	7219 516180	Parcu
2)	भाग्ने भगव्यात	चडिमार्स्डिंद	9-8-	749876 7061	Aldas
3)	satish Hanmant Yadav	obad - Kavathe T-Karad	10-8-2019	95617550	Brada/
4)	थाहत दशाख उने क	कडेपूर ता कड़ेगव जि. भागती	10-8-2019	88888033FP	£34-2149
5]	Mankar Jalindor Sadhy	A.P. Yedemachindra,	5-12-2019	7058231506	#J.SM
6]	Dipak Shivayirao Mali	A.P. Pethnaka.	6-2-2020		
7)	Makesh Pralhad kale	A. P. Maldhan	92-2-2020	ee 600 9 3 e380	Kalump.
8)	जाबव मंनय सळाशम	मुःयो सामवडे	92-02-2020	92053630 92053630	H. 7. 912 4
	Head	Chavan College			
Y	Department of Chemistry ashwantrap Chavan College of Science, Karad	CHEMISTRY	of Scin		
		* Karad * and			

Department of Chemistry

Date: 30.8.2019

Name of Activity-To determine percentage of Nitrogen in the Soil and pH of Soil.

pH values - '5

Blank Titration

Back Titration

% of Nitrogen in the given sample = $\frac{(X-Y)*N* \ 0.14}{Amount of soil sample \ in \ gm}$

X= Volume of 0.1 N HCL required (Blank reading)

Y = Volume of 0.1 N HCL required (sample reading)

N= Normality of HCl

$$= (18.00 - 25.00) \times 0.1 \times 14$$

conclusion- The soil contain 1.967. Ni mogen 4 notyse q soil is audic.



Department of Chemistry

Date: 30 8.2019

Name of Activity -To determine percentage of Nitrogen in the Soil and pH of Soil.

Name of Students- Rokade Prhyanka Vinayale

Islampur, Id-Valva, Dist. - Sangli Address -

% of Nitrogen in Soil is: 1-98-/.
Nature of Soil is: 5 Ph _ aidic

Department of Chemistry

Date: 30.8.2019

Name of Activity-To determine percentage of Nitrogen in the Soil and pH of Soil.

Name of Students- Azium Jagamnath Lohar
Address- Yedemachindra Tul. Valva, Dist. Bungali

pH values -

Blank Titration

Back Titration

20 ml

16 m

2. 22 m

2. 17 ml

3. 20 ml

3. 16 ml

CBR = 20 ml

CBR = 16 ml

% of Nitrogen in the given sample = $\frac{(X-Y)*N*0.14}{Amount of soil sample in gm}$

X= Volume of 0.1 N HCL required (Blank reading)

Y = Volume of 0.1 N HCL required (sample reading)

N= Normality of HCl

N=0.1 N

X = 20 ml 1/2 g Nimogen in = [Y-X] x N x 14 Y = 16 ml the given sample Amount q soil sample

= [16-20] x 6.1 X14

= 2.8.1.

Conclusion - This sou contain 1-96%. Nimogen 4 nouture 9 soil is hasic.



Department of Chemistry

Date:

30.8.2019

Name of Activity -To determine percentage of Nitrogen in the Soil and pH of Soil.

Name of Students- Arjun Jagamoth Lohar

Address- A/P- Yedemachindra, Tal-Valva, Dist-Sangli

% of Nitrogen in Soil is: 2.8 1.

Nature of Soil is

: 8PH basic

Department of Chemistry

Date: 80.8.2019

Name of Activity-To determine percentage of Nitrogen in the Soil and pH of Soil.

Name of Students- Batish Hanmaut Yadav

Address - A/P - Kavathe, Tal - Karad, Dist Satara

pH values -

Blank Titration

Back Titration

1. 25.00 ml

1. 18.00 m

2. 26.00 m

2. 19.00 m

3. 25.00 m

3. 18.00 W

CBR = RJ-10 ml

CBR = 18.00 W

% of Nitrogen in the given sample = $\frac{(X-Y)*N*0.14}{Amount of soil sample in gm}$

X= Volume of 0.1 N HCL required (Blank reading)

Y = Volume of 0.1 N HCL required (sample reading)

N= Normality of HCl

$$X = 25 \text{ m}$$
 $Y = 1.8 \text{ m}$
 $Y = 1.8 \text{ m}$
 $Y = 0.1 \text{ N}$

Sample

Sample

 $Y = 1.8 \text{ m}$
 $Y = 0.1 \text{ N}$

Sample

 $Y = 1.8 \text{ m}$
 $Y = 0.1 \text{ N}$

Sample

 $Y = 1.8 \text{ m}$
 $Y = 1.8 \text{ m}$

Sample

 $Y = 1.8 \text{ m}$
 $Y = 1.8 \text{ m}$

Sample

 $Y = 1.8 \text{ m}$

Sample

Conclusion- This soil contain 1.96%. Nitrogen & nouture que sail is audie.



Department of Chemistry

Date: 80.8.2019

Name of Activity -To determine percentage of Nitrogen in the Soil and pH of Soil.

Name of Students- Yadav Satish Hanmant

Address - A/P- Kavathe, Tal- Rasad, Pist. - Salasa

% of Nitrogen in Soil is: 1.96 1/.

Nature of Soil is : 6

Chavan College of Scientific CHEMISTRY CHEMISTRY

Department of Chemistry

Date: 30 · 8 · 2019

Name of Activity-To determine percentage of Nitrogen in the Soil and pH of Soil.

Name of Students- Dashruth Anty Tadav

pH values -

Blank Titration

% of Nitrogen in the given sample = $\frac{(X-Y)*N*0.14}{Amount of soil sample in gm}$

X= Volume of 0.1 N HCL required (Blank reading)

Y = Volume of 0.1 N HCL required (sample reading)

N= Normality of HCl

$$X=20 \text{ ml}$$
 ? $Y=16 \text{ me}$ O. Ly. 9 no mogen in the $Y=16 \text{ me}$ 9 of given sample = $[Y-X] \times N \times 14$

Amount g 800 sample in gm.

= $[16-20] \times 0.1 \times 14$

= 2.8% .

Inclusion - This 801 contain 1-96%. No mogen f nature $[X-Y] \times [X-Y] \times [X-$

Conclusion -2 soil is basic.



Department of Chemistry

Date: 3 0 .8 . 2019

Name of Activity -To determine percentage of Nitrogen in the Soil and pH of Soil.

Name of Students- Dash roth Antu Yadav

Address- Mp- kadepur, Tel- kadegaan, Dist. - Sangali

% of Nitrogen in Soil is: 2.8 1.

Nature of Soil is : 8 [basic]

CHEMISTRY OF Karad * 20

Department of Chemistry

Date: 3/02/2020

Name of Activity-To determine percentage of Nitrogen in the Soil and pH of Soil.

Mankar Jalindar Sadhu Name of Students-

Address - A.P. Yedemachindra

pH values -

Blank Titration

25 m

2. 24.5 m

3. 25 ml

CBR = 25 ml

Back Titration

20 ml

2. 18 m

20 ml

CBR = 20 ml

% of Nitrogen in the given sample = $\frac{(A-I)*N*0.1*}{Amount of soil sample in gm}$

X= Volume of 0.1 N HCL required (Blank reading)

Y = Volume of 0.1 N HCL required (sample reading)

N= Normality of HCl

7. 9 Nitrogen on the given sample
$$= \frac{(y-x) \times N \times 14}{amount q}$$
 soil

 $= (25-20) \times 0.1 \times 14$
 $= 1.4.7.$



Department of Chemistry

Date: 3/02/2020

Name of Activity -To determine percentage of Nitrogen in the Soil and pH of Soil.

Name of Students- Maukar Julindar Sadhu

Address - A. P. Medemachindra

% of Nitrogen in Soil is: 1.4 %.

Nature of Soil is : a call

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Department of Chemistry

Date: 3/02/2020

Name of Activity-To determine percentage of Nitrogen in the Soil and pH of Soil.

pH values -

Blank Titration

20 m

22 ml

20 ml

CBR = 20 ml

Back Titration

18 mf

2. 20 ml 3. 19 ml

CBR = [9.5 m]

% of Nitrogen in the given sample = $\frac{(X-Y)*N*0.14}{Amount of soil sample in gm}$

X= Volume of 0.1 N HCL required (Blank reading)

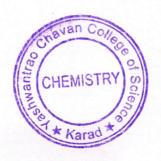
Y = Volume of 0.1 N HCL required (sample reading)

N= Normality of HCl

7. 9 Nimogen on the given sumple =
$$(y-x) \times N \times 14$$

amount q soil sample in gm

= $(20-19.5) \times 0.1 \times 14$
 $= 0.14 \times 14$



Department of Chemistry

Date: 3 | 02/2020

Name of Activity -To determine percentage of Nitrogen in the Soil and pH of Soil.

Name of Students-Pipak Shavajirao Mahaddress- A'P. Pethnaka

% of Nitrogen in Soil is: 0.14 '/・

Nature of Soil is

: Basic

Department of Chemistry Yashwantrao Chavan College of

Science, Karad

Department of Chemistry

Date: 3 1 0 2 2020

Name of Activity-To determine percentage of Nitrogen in the Soil and pH of Soil.

Name of Students- Mahesh Prachad Kale

Address - A.P. Maldhan

pH values - q

Blank Titration

1. 22 ml

2. 22 ml

3. 22 ml

CBR = 22 ml

Back Titration

1 somt

2 19 md

3. 20 ml

CBR = 20 ml

% of Nitrogen in the given sample = $\frac{(X-Y)*N* \ 0.14}{Amount of soil sample in gm}$

X= Volume of 0.1 N HCL required (Blank reading)

Y = Volume of 0.1 N HCL required (sample reading)

N= Normality of HCl

7. 9 Nitrogen on the given sample =
$$(Y-2e) \times N \times 14$$

= $(22-20) \times 0.1 \times 14$
5
= 0.564 .



Department of Chemistry

Date: 3/02/26 20

Name of Activity -To determine percentage of Nitrogen in the Soil and pH of Soil.

Name of Students- Mahesh Pralhad Kode

Address -A.P. Maldhan

% of Nitrogen in Soil is:

0.567.

Nature of Soil is

Basic

Yashwantrao Chavan College of Science, Karad

Extension Activity 2019-20

Department of Chemistry

Date: 15/04/2020

Name of Activity-To determine percentage of Nitrogen in the Soil and pH of Soil.

Name of Students- Jachar Lanjay Sakharam

A.P. Tasardi

pH values -

Back Titration Blank Titration

75 m

18 m

24.5 m

25 m

3. 20 ml

CBR = 25 ml

CBR = 20 m

% of Nitrogen in the given sample = $\frac{(X-Y)*N*0.14}{Amount of soil sample in gm}$

X= Volume of 0.1 N HCL required (Blank reading)

Y = Volume of 0.1 N HCL required (sample reading)

N= Normality of HCl

N= Normality of HCI

7. 9 the Nitrogen in the given sample = $(y-2) \times N \times 14$ amount g soil

sample in gm

= $(25-20) \times 0.1 \times 14$ = 1.4.1.

Conclusion -1.4.1. Nitrogen & This Soil Contains nature a soil is audic.



Department of Chemistry

Date: 15/04/2020

Name of Activity -To determine percentage of Nitrogen in the Soil and pH of Soil.

Name of Students- Jadhar Banjay sakharam Address- A.P. Tagardi

% of Nitrogen in Soil is: 1.4 1/.

Nature of Soil is : aidic